Review Session Problems 2

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Exercise 1. When we solve the hydrogen atom, we assume that the nucleus is a point charge. In this problem, we will compute the approximate change to the energy levels due to the finite size of the nucleus. This is called the **volume effect**. Model the nucleus as a uniform sphere of radius $r_0 A^{1/3}$, where $A^{1/3}$ is the number of nucleons (so this works for e.g. deuterium) and $r_0 = 1.3 \cdot 10^{-13}$ cm.

- a) What is the potential V(r)?

 Hint: Outside the nucleus, V(r) is just the Coulomb potential. Inside the nucleus, use Gauss' law to determine V(r).
- b) What is H', where H^0 is the hydrogen atom hamiltonian?
- c) It turns out that the $\ell=0$ states are the most affected by this perturbation, with the $\ell>0$ states only very slightly affected. Calculate the correction to the energy levels for all states with $\ell=0$. Note that

$$R_{n0}(0) = \frac{2}{(na_0)^{3/2}},$$

where $a_0 = \hbar^2/me^2$.

- d) For hydrogen, calculate the correction to the n = 1 and n = 2 states in eV.
- e) Fine structure is of order $\alpha^4 mc^2$. Compare the magnitude of the volume effect to that of fine structure.

Exercise 2. Explain the physical origins of

- a) fine structure
- b) Lamb shift
- c) hyperfine structure.

Exercise 3. Griffiths 8.19 Find the lowest bound on the ground state of hydrogen using the variational principle and an exponential trial wavefunction,

$$\psi(\mathbf{r}) = Ae^{-br^2}.$$

where A is determined by normalization and b is a variational parameter. Express your answer in eV.

Exercise 4. *Griffiths 9.18* When we turn on an external electric field, it should be possible to ionize the electron in an atom. A crude model for this is to suppose that a particle is in a very deep, one-dimensional finite square well.

- a) What is the energy of the ground state, measured up from the bottom of the well? Assume that $V_0 \gg \hbar^2/ma^2$.
- b) Introduce the perturbation $H' = -\alpha x$, where $\alpha \equiv eE_{\rm ext}$. Assume that $\alpha a \ll \hbar^2/ma^2$, and sketch the total potential, noting that the electron can tunnel out in the direction of positive x.
- c) Calculate

$$\gamma = \frac{1}{\hbar} \int |p(x)| \, \mathrm{d}x,$$

and estimate the time it would take for the particle to escape,

$$\tau = \frac{2x_1}{v}e^{2\gamma},$$

where x_1 is the distance the electron must travel to reach the tipping point of the potential and v is the speed of the electron.

d) Plug in some numbers, e.g. $V_0 = 20$ eV, $E_{\rm ext} = 7 \cdot 10^6$ V/m, $a = 10^{-10}$ m. Calculate τ , and compare it to the age of the universe.